## Communications to the editor

## ASPARENOMYCIN A, A NEW CARBAPENEM ANTIBIOTIC

Sir:

In the course of our screening work for new  $\beta$ -lactam antibiotics, we found that a streptomycete numbered PA-31088, which was identified as a new species and named *Streptomyces tokunonensis* sp. nov., produces a new carbapenem antibiotic named asparenomycin A\* and some related antibiotics. These antibiotics were also found in the culture broth of another strain PA-39504 identified as *Streptomyces argenteolus*.

The strain PA-31088 was fermented in 30-liter jar fermentor containing 20 liters of a medium composed of 2.4% tomato paste, 2.4% dextrin, 1.2% dry yeast, 0.0006% CoCl<sub>2</sub>·6H<sub>2</sub>O, at 28°C under agitation of 300 rpm and aeration of 20 liters per minute for 65 hours.

The antibiotics in the culture filtrate (100 liters) were adsorbed on a column of Amberlite IRA-68 (Cl<sup>-</sup>) (Rohm and Haas Co., Ltd.) and eluted with 5% NaCl. The active eluate was desalted on a Diaion HP-20 (Mitsubishi Kasei Kogyo Co., Ltd.) column and then adsorbed on an activated carbon at pH 5.0 and eluted with 60% acetone (pH 7.0). Evaporation and lyophilization of the eluate gave a crude powder (20 g). Isolation of asparenomycin A was achieved by successive column chromatography on a Pre PAK-500/C<sub>18</sub> column of a High Speed Liquid Chromatography System 500 (Waters Co., Ltd.) and a column of a Diaion HP-20AG (200~ 400 mesh) with phosphate buffer solution, pH 7.0. The fraction of asparenomycin A from the last column was desalted on an HP-20 column and lyophilized. A substantially pure preparation of the antibiotic was obtained as the sodium salt (210 mg).

Asparenomycin A sodium salt is a colorless amorphous powder, which gradually decomposes above 150°C, and is soluble in water, methanol and dimethylsulfoxide but insoluble in ethyl acetate, acetone and chloroform. It shows positive color reaction to EHRLICH's reagent but is ninhydrin-negative. On paper electrophoresis in

\* Presented in Japan Kokai (patent) 55-13,628 (Aug. 10, 1980) as a name PA-31088-IV. 50 mM phosphate buffer, pH 7.0 at 10 volt/cm for 3 hours, the antibiotic moved towards the anode with the same mobility as that of penicillin N. Elemental analysis (Found: C, 45.30; H, 4.82; N, 7.74; S, 7.86; Na, 5.99) and FD mass: 355 (MH<sup>+</sup> of the methyl ester) agreed with a molecular formula  $C_{14}H_{16}N_2O_6S$  for the free acid. The antibiotic shows UV (10 mM phosphate buffer, pH 7.0),  $\lambda_{max}$ : 241 nm ( $\varepsilon$ , 21472), 280 nm

Table 1. <sup>1</sup>H NMR spectra of I and II sodium salts.

Assignment	$\delta$ ppm (J Hz)		
	I	II	
9-CH <sub>3</sub>	1.99 (s)	1.85 (s)	
17-CH <sub>3</sub>	2.12 (s)	2.16 (s)	
		2.17 (s)	
$4-CH_2$	3.16 (d-like)	2.0~2.9 (m)	
5-CH	5.01 (t-like)	$4.6 \sim 5.4$ (m)	
3-CH			
10-CH <sub>2</sub>	4.26 (s)	4.23 (s)	
		4.24 (s)	
13-CH	6.34 (d, 14.0)	6.11 (d, 14.0)	
		6.22 (d, 14.1)	
14-CH	7.53 (d, 14.0)	7.50 (d, 14.0)	
		7.61 (d, 14.1)	

Spectra were recorded with a Varian XL-100-12A spectrometer in  $D_2O$  at room temperature using DSS as an internal reference.

Table 2. <sup>13</sup>C NMR spectrum of I sodium salt.

Assignment (Type)	δ ppm	Assignment (Type)	δ ppm
9 (-CH <sub>3</sub> )	15.9 (q)	14 (=CH-)	134.7 (d)
17 (-CH <sub>3</sub> )	23.0 (q)	$3^{a} (=C \leq)$	135.6 (s)
4 (-CH <sub>2</sub> -)	32.5 (t)	$2^{a} (=C \leq)$	143.2 (s)
5 (>CH-)	60.4 (d)	8 (=C<)	150.3 (s)
10 (-CH <sub>2</sub> -)	64.5 (t)	7 (O=C<)	166.7 (s)
13 (=CH-)	111.7 (d)	16 ( $\mathbf{O} = \mathbf{C} \langle \rangle$ )	173.2 (s)
$6^{a} (=C\langle)$	134.2 (s)	11 ( <b>O</b> = <b>C</b> <)	174.6 (s)

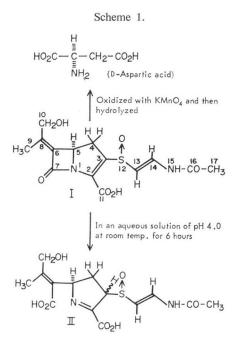
Spectrum was recorded with a Varian XL-100-12A spectrometer in D<sub>2</sub>O at 5°C using CH<sub>3</sub>CN as an internal reference.  $\delta$ : Calculated by assuming  $\delta$  (CH<sub>3</sub>CN)=1.7 ppm from DSS.

<sup>a</sup> Tentative assignments.

(shoulder), 320 nm (shoulder); IR (KBr), 1750, 1695, 1620, 1380, 1270, 1200, 1045, 950 cm<sup>-1</sup>;  $[\alpha]_{2^{\circ}}^{\circ\circ} -210.8 \pm 5.1^{\circ}$  (*c* 0.536, 10 mM phosphate buffer, pH 7.0); CD:  $[\theta]_{390}$  0,  $[\theta]_{315} -22754$ ,  $[\theta]_{278} -102702$ ,  $[\theta]_{258.5}$  0,  $[\theta]_{243} +93292$ ,  $[\theta]_{100} +14480$  (*c* 0.0458, 10 mM phosphate buffer, pH 7.0). The signals and assignments of <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra are shown in Tables 1 and 2.

Interpretation of these physico-chemical properties and chemical behaviors described below lead to the structure of asparenomycin A as I in Scheme 1.

In an aqueous solution of pH 4.0, asparenomycin A was hydrolyzed to a non  $\beta$ -lactam compound (II) (the absorption at  $1750 \text{ cm}^{-1} \text{ dis-}$ appeared) at room temperature within 6 hours. Most peaks of the <sup>1</sup>H NMR spectrum of II were observed as pairs, suggesting that  $\Pi$  is a mixture of diastereoisomers caused by appearance of a new additional proton (Table 1).<sup>1)</sup> II moves two times faster than penicillin N in the paper electrophoresis. II shows a positive EHRLICH reaction but a negative ninhydrin reaction. When an aqueous solution of II of pH 2.0 was allowed to stand at room temperature for 16 hours, no significant changes were detected by thin-layer chromatography and by paper electrophoresis. A high possibility of E orientation at C-8 was indicated by the fact that II does not give detectable  $\gamma$ -lactone. When asparenomycin A was



oxidized with KMnO<sub>4</sub> and then hydrolyzed with 6 N HCl at 110°C for 2 hours, a significant amount of aspartic acid was detected by an automatic amino acid analyzer. The configuration of the amino acid was determined to have D-configuration from the result of HPLC of its L-leucylated product. Consequently the *R* configuration at 5-C is concluded.

Organism	MIC (µg/ml) <sup>a</sup>			
organism	Asparenomycin A	Ampicillin	Cefoxitin	
Staphylococcus aureus 209P JC-1	1.56	0.1	1.56	
Staphylococcus aureus C-14 <sup>b</sup>	1.56 6.25		3.13	
Streptococcus pyogenes C-203	1.56 0.05		0.78	
Escherichia coli NIHJ JC-2	1.56	6.25	3.13	
Escherichia coli 377°	0.39	100	12.5	
Klebsiella pneumoniae SRL-1	0.78	0.78	1.56	
Klebsiella sp. 363 <sup>b</sup>	0.78	>100	1.56	
Proteus mirabilis PR-4	3.13	1.56	1.56	
Proteus vulgaris CN-329	12.5	50	3.13	
Enterobacter cloacae 233	1.56	50	>100	
Serratia marcescens ATCC 13880	12.5	25	12.5	
Pseudomonas aeruginosa ATCC 25619	25	>100	>100	

Table 3. Antibacterial activity of asparenomycin A, ampicillin and cefoxitin.

<sup>a</sup> Determined by agar dilution method in MUELLER-HINTON agar and inoculated by one loopful of ca. 10<sup>6</sup> cells per ml.

<sup>b</sup> Penicillinase producing strain.

<sup>c</sup> Cephalosporinase producing strain.

Table 4. Inhibition of  $\beta$ -lactamases produced by Gram-negative bacteria by asparenomycin and clavulanic acid.

Source of <i>B</i> -lactamase <sup>a</sup>	Class <sup>b</sup>	Minimum effective concentration (µg/ml)°	
Source of p-ractamase"		Asparenomycin A	Clavulanic acid
Escherichia coli 6	Ib	1.0	250
Enterobacter cloacae 92	Ia	1.0	>250
Proteus vulgaris 31	Ic	0.001	1
Escherichia coli W3110 RTEM	IIIa	0.016	0.063
Klebsiella sp. 363	IV	0.063	0.063
Enterobacter cloacae 53	IVa	0.25	0.063

<sup>a</sup> Enzyme preparations used were partially purified.

<sup>b</sup> Classification of RICHMOND and SYKES.

<sup>c</sup> Inhibitor was incubated with enzyme at 30°C for 10 minutes prior to adding nitrocefin (50  $\mu$ g/ml) and minimum concentration to inhibit color change was determined.

The complete structure of asparenomycin A including the orientation of sulfoxide (R) was confirmed by X-ray crystallographic analysis<sup>2</sup>) of the *p*-nitrobenzyl ester and the methyl ester of the antibiotic.

Asparenomycin A is unique in having the substituted ethylidene side chain in the structure and is obviously distinguished from hitherto known carbapenem antibiotics such as thienamycin, olivanic acid derivatives, PS-5 and carpetimycins A and B.<sup>3)</sup>

Asparenomycin A is active against a broad range of Gram-positive and Gram-negative bacteria including  $\beta$ -lactamase producing organisms as shown in Table 3. It shows strong inhibitory activity against various type of  $\beta$ -lactamases as shown in Table 4.

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